

# Chemical Equations Gizmo Answers Online

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Balancing Chemical Equations - Answer Key Balance the equations below: 1)  $1 \text{ N}_2 + 3 \text{ H}_2$

### **Balancing Chemical Equations Answer Key**

Balancing Chemical Equations -Answer Key Balance the equations below: 1)  $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$  2)  $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$  3)  $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$  4)  $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$  5)  $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$  6)  $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2(\text{SO}_4)_3$

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Balancing Equations Worksheet and Key 1 Answer the following questions about the chemical equation shown below:  $2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$  a) What are the reactants? b) What is the product? c) What do we call the number "2" in front of the  $\text{H}_2$  (and  $\text{H}_2\text{O}$ )? d) Is the reaction balanced? e) Why is there not a coefficient for  $\text{O}_2$ ?

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